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4.1 General Properties of Aqueous Solutions •

A solution is a homogeneous mixture of two or more substances. • A solution is made when

one substance (the solute) is dissolved in another (the solvent). • The solute is the

substance that is present in ...

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2. Placing aluminum foil in any solution

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containing aqueous copper(II) ions will result in a reaction. The reaction is slow to begin with, then proceeds rapidly. a) Referring to a solubility table, name at least four ionic compounds that could be dissolved in water to make a solution containing aqueous copper(II) ions.

AP Chemistry - Molecular, Ionic and Net Ionic Equations

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